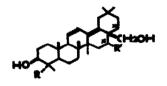
Tetrahedron Letters No.39, pp. 4721-4724, 1956. Pergamon Press Ltd. Printed in Great Britain.

SAIKOGENIN E, A GENUINE SAPOGENIN OF BUPLEURUM ROOTS

Faculty of Pharmaceutical Sciences, University of Tokyo, Japan

(Received 4 July 1966; in revised form 23 July 1966)

Previously we reported (1) on the structure of saikogenin A, which was obtained as a sapogenin of saikoside Ia, a saponin of the roots of <u>Bupleurum falcatum</u> L.(** Δ). An only ambiguous position of a secondary hydroxyl (16β or 22β) in our formula of saikogenin A has been proved to be located at 16β by Kubota et al.(2) shortly after the publication of our report. The complete structure of saikogenin A has been illustrated by Kubota et al. in relation with other sapogenins, saikogenins C, D, and B, which were isolated by them along with longispinogenin from the hydrolysate of Bupleurum saponin.



но СНЗОН

- (I) Saikogenin A R=CH₂OH, R'= =OH
- (IV) Saikogenin B
- (II) Saikogenin C R=CH3, R'==OH
- (III)Saikogenin D R=CH_OH, R'= ---OH

The β -configuration of H at $C_{(18)}$ was indicated by Dr.T.Kubota in his private communication.

4722 No. 39

The purified saikoside II, one of the main saponius of <u>Rupleurum</u> roots, showed no remarkable UV- absorption above 210 mm, whereas saikogenin C, m.n. 301° which was obtained by the acid hydrolysis of saikoside II gave UV-absorptions of heteroannular diene at 241, 250 and 260 mm. It would, therefore, be reasonable to assume that saikogenin C would be an artifact. The present study deals with a genuine sapogenin isolated from saikoside II fraction, which has now been designated saikogenin E.

Saikogenin E was obtained from the sabonin by the Smith degradation(3) carried out as follows: The sugar moiety of saikoside II was cleaved by the oxidation with periodate, and the aldehydic product was reduced with NaBH₄ to an alcoholic compound which was hydrolysed under a mild condition using 0.1 n $\rm H_2SO_4$ in methanol at room temperature to afford an analyzone (saikogenin E). The same analyzone was also formed by heating the aldehydic intermediate with 5% KCT.

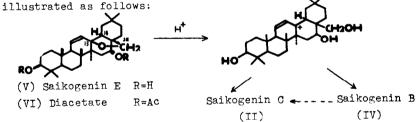
Saikomenin E (V), $C_{30}H_{48}O_3$, m.p. 289° (decomp.), $C_{30}C_{10}^{20}$ + 112° (c=0.5 in CUCl₃) shows no UV-absorption above 210 mu indicating the absence of conjugated double bonds. On heating V with 50° $V_{2}SO_{4}$ for 30 mins. on a boiling water bath, saikogenin C (II) was afforded. As the mother liquour of saikogenin C showed UV absorption at 281 mµ (homoannular diene), saikogenin 3 was also litely formed from saikogenin E by the action of acid. On acetylation with acetic anhydride and pyridine at room temperature, V yielded a diacetate (VI), $C_{34}H_{52}O_5$, m.p. 213°,

Sailrogenin C was identified with Kubota's specimen by a mixed fusion and comparison of IR spectra and thin layer chromatograms.

No.39 4723

 $\text{LaZ}_{\text{D}}^{23}$ + 117.7° (c=1.1 in CHCl₃), which revealed the absence of unblocked hydroxyl in the IR spectrum and the nonexistence of conjugated double bonds by the UV-absorption. The n.m.r.spectrum (measured at 100 Mc in $\mathrm{CDCl}_{\mathsf{X}}$) of V indicated the presence of 7 tertiary methyls (τ 8.91~9.23), a methylene adjacent to an ether linkage (7 6.90 and 6.11 coupling in AB type, J=7.7 c.p.s.) and olefinic protons (τ 4.59 and 4.17 coupling in AB type, J= 11.5 c.p.s. with small doubled coupling in the former signal, J = 3 c.p.s.). The n.m.r. spectrum (measured at 60 Mc in CDCl₃) of VI showed 2 acetyl groupings (7 8.01 (s)) in addition to the presence of the same systems as shown in V (7 6.84 and 6.07 coupling in AB type (J = 7.5 c.p.s.) and 74.65 and 4.18 coupling in AB type (J = 10.5 c.p.s.)). The signals at 76.78 (m) and 5.78 (m) in the n.m.r. apectrum of V shifted to 5.60 (m) and 4.60 (m) in that of VI , respectively. It would indicate that the hydroxyls of V are secondary locating at 38 and 168, in the correlation with saikogenin C.

The above mentioned results led to a conclusion that the ether linkage in saikogenin E must be located between C₍₂₈₎ and a tertiary carbon atom at the position 13. Saikogenin E has now been formulated as (V), and the formation of saikogenins C and B is



4724 No.39

From the biosynthetical point of view, the configuration at $C_{(18)}$ is likely to be β since longispinogenin is occurring in the same plant.(2)

Acknowledgements: We are grateful to Dr. T. Kubota, Shionogi Research Laboratory for identification of saikogenins B and C. We are also indebted to Miss Y. Shibanuma, National Institute of Radiological Science and Miss M. Ohnishi, National Cancer Center Research Institute for the measurement of n.m.r. spectra. Thanks are also due to Yakurikenkyukai for grant, and to Research Laboratory of Takeda Chemical Industries Ltd. for supplying plant material.

REFERENCES

- 1. S. Shibata, I. Kitagawa and H. Fujimoto: <u>Tetrahedron Letters</u>, No. 42, 3783 (1965); <u>Chem. Pharm.Bull.(Tokyo),14(1966)</u>in press.
- 2. T. Kubota, F. Tonami and H. Hinoh: <u>Tetrahedron Letters</u>, No.7, 701 (1966).
- 3. F. Smith et al. "Methods in Carbohydrate Chemistry" (Ed. R.L. Whistler) Vol.V, p.361 (Academic Press, New York & London, 1965).